

ORGANIC CHEMISTRY PAULA YURKANIS BRUCE 6TH EDITION FULL



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Essential Organic Chemistry THIRD EDITION Paula Yurkanis Bruice UNIVERSITY OF CALIFORNIA SANTA BARBARA 37716_00_FM_pi-xxiv.indd 1 11/21/14 1:28 PM

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In chemistry, specific rotation ($[\alpha]$) is a property of a chiral chemical compound.: 244 It is defined as the change in orientation of monochromatic plane-polarized light, per unit distance–concentration product, as the light passes through a sample of a compound in solution.: 2–65 Compounds which rotate the plane of polarization of a beam of plane polarized light clockwise are said to be ...

Specific rotation - Wikipedia

Alkener er en hydrokarbon-familie, som består av hydrogen og karbon. Alkener ligner alkaner, men alkener har minst en dobbeltbinding mellom to karbonatomer.. De tre minste alkenene er: eten, propen og buten. Alkener må ha minst to karbonatomer, meten eksisteres derfor ikke. Formelen for alkener er: $C_n H_{2n}$ eksempel: penten er: $C_5 H_{10}$. Umattede hydrokarboner som alkener kan påvises ved å ...

Alkener – Wikipedia

Alkali, også kalt baser, er det eldre navnet på ammoniakk og oksider, karbonater av alkalimetaller (natrium, kalium, litium m.m.) og hydroksider. Alkali kan gi alkalisk reaksjon i forbindelse med vann. Navnet alkali kommer antakelig av at aske i vann gir basisk løsning (alkali er arabisk og betyr aske).

Alkali – Wikipedia

In chemistry, orbital hybridisation (or hybridization) is the concept of mixing atomic orbitals into new hybrid orbitals (with different energies, shapes, etc., than the component atomic orbitals) suitable for the pairing of electrons to form chemical bonds in valence bond theory. Hybrid orbitals are very useful in the explanation of molecular geometry and atomic bonding properties and are ...

Orbital hybridisation - Wikipedia

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Die Markownikow-Regel (nach Wladimir Wassiljewitsch Markownikow) beschreibt in der organischen Chemie die Produkte einer elektrophilen Addition von Halogenwasserstoffen an Kohlenstoff-Kohlenstoff-Doppelbindungen in unsymmetrischen Alkenen.. Die Regel besagt, dass bei der Anlagerung von Halogenwasserstoffen an asymmetrische Alkene das Wasserstoffatom immer an das bereits wasserstoffreichere ...

Markownikow-Regel – Wikipedia

Sauerstoff ist elektronegativer als Wasserstoff und Kohlenstoff, d. h., er zieht Elektronen stärker an als diese. Das führt zu einer unsymmetrischen Verteilung der Elektronen entlang der C–O–H-Bindung, man spricht von einer polaren Bindung, es bildet sich ein molekularer Dipol aus. Diese Dipole können untereinander Wasserstoffbrückenbindungen ausbilden, die die Anziehung der einzelnen ...

Alkohole – Wikipedia

En chimie, l'équation de Henderson-Hasselbalch est une équation donnant le pH d'un système tamponné. Elle est notamment utilisée en physiologie et en médecine pour déterminer le pH sanguin à partir des concentrations en ion bicarbonate et en acide carbonique

Équation de Henderson-Hasselbalch — Wikipédia

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Mechanika kwantowa. Zagadnienie p?askiej budowy pier?cienia benzenowego dobrze t?umaczy kwantowomechaniczna metoda orbitali molekularnych. Przyjmuje si? hybrydyzacji sp² (k?ty pomi?dzy wi?zaniem H-C-C i C-C-C wynoszą 120°) dla atomów w?gla. Zhybrydyzowane orbitale tworzą wi?zania ? C-C i C-H. Pozosta?e orbitale p atomów w?gla (o osiach prostopad?ych do p?aszczyzny ...

Benzen – Wikipedia, wolna encyklopedia

Aminohapot ovat orgaanisia yhdisteitä, joissa on sekä amino- (-NH₂) että karboksyyliiryhmä (-COOH) liittyneenä samaan molekyyliin. Aminoryhmä on emäksinen ja karboksyyliiryhmä hapan. Aminohapot polymeroituvat polypeptideiksi. Polypeptidiketjun pituuden ylittäessä 50 sitä aletaan yleensä kutsua proteiiniksi. Ihmiselle välttämättömiä aminohappoja on 20, joista 11 elimistö osaa ...

Aminohapot – Wikipedia

Hybrid orbital
hybridization ...